

# Effects of past, present, and future ocean carbon dioxide concentrations on the growth and survival of larval shellfish

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The combustion of fossil fuels has enriched levels of CO<sub>2</sub> in the world's oceans and decreased ocean pH. Although the continuation of these processes may alter the growth, survival, and diversity of marine organisms that synthesize CaCO<sub>3</sub> shells, the effects of ocean acidification since the dawn of the industrial revolution are not clear. Here we present experiments that examined the effects of the ocean's past, present, and future (21st and 22nd centuries) CO<sub>2</sub> concentrations on the growth, survival, and condition of larvae of two species of commercially and ecologically valuable bivalve shellfish (*Mercenaria mercenaria* and *Argopecten irradians*). Larvae grown under near preindustrial CO<sub>2</sub> concentrations (250 ppm) displayed significantly faster growth and metamorphosis as well as higher survival and lipid accumulation rates compared with individuals reared under modern day CO<sub>2</sub> levels. Bivalves grown under near preindustrial CO<sub>2</sub> levels displayed thicker, more robust shells than individuals grown at present CO<sub>2</sub> concentrations, whereas bivalves exposed to CO<sub>2</sub> levels expected later this century had shells that were malformed and eroded. These results suggest that the ocean acidification that has occurred during the past two centuries may be inhibiting the development and survival of larval shellfish and contributing to global declines of some bivalve populations.

bivalve larvae | climate change | ocean acidification

More than 8 Pg of carbon dioxide (CO<sub>2</sub>) is released annually into our planet's atmosphere via the combustion of fossil fuels (1). About one-third of anthropogenically derived CO<sub>2</sub> has entered the world's oceans during the past two centuries (2) and atmospheric and surface ocean CO<sub>2</sub> levels are expected to reach ~750 ppm by 2100 (3, 4). CO<sub>2</sub> entering the ocean decreases the availability of carbonate ions (CO<sub>3</sub><sup>2-</sup>) and reduces ocean pH, a process known as ocean acidification. These changes in ocean chemistry may have dire consequences for ocean animals that produce hard parts made from calcium carbonate (CaCO<sub>3</sub>). The experimental enrichment of CO<sub>2</sub> to levels expected in the coming century has been shown to dramatically alter the growth, survival, and morphology of numerous calcifying organisms including coccolithophores, coral reefs, crustose coralline algae, echinoderms, foraminifera, and pteropods (5–7). Many shellfish also produce calcareous shells, and juvenile and adult clams, mussels, and oysters have been shown to be adversely affected by elevated CO<sub>2</sub> (8–12). The earliest life history stages of shellfish, larvae, have been shown to be especially vulnerable to high CO<sub>2</sub>, displaying large declines in survival and delays in metamorphosis at levels predicted to occur later this century, suggesting recruitment of these populations may be adversely impacted by ocean acidification (12–14).

Although it is clear that calcifying ocean animals such as shellfish are sensitive to the increases in CO<sub>2</sub> projected for the future, the extent to which the rise in CO<sub>2</sub> that has occurred since the dawn of the industrial revolution has impacted these populations is poorly understood. Here we present experiments that examined the effects of past (250 ppm), present (390 ppm), and future (>400 ppm) CO<sub>2</sub> concentrations on larvae of two species of shellfish: the Northern quahog or hard clam, *Mercenaria mercenaria*, and the bay scallop, *Argopecten irradians*. These bivalves are ecologically and

commercially valuable resources: US mollusk harvests are \$750 million annually (15), with ecosystem services far exceeding that value (16, 17). For experiments, CO<sub>2</sub> was delivered via a gas proportionator system and CO<sub>2</sub> levels in seawater were determined by quantifying dissolved inorganic carbon and pH during experiments using an EGM-4 Environmental Gas Analyzer (PP Systems) and the program CO2SYS (<http://cdiac.ornl.gov/ftp/co2sys/>). Dissolved inorganic carbon was measured with a methodological precision of ±3.6% and full recovery (102 ± 3%) of Dr. Andrew Dickson's (Scripps Institution of Oceanography, University of California at San Diego, La Jolla, CA) certified reference material for total inorganic carbon in seawater [Batch 102 = 2,013 μmol dissolved inorganic carbon (DIC) kg seawater<sup>-1</sup>] was obtained with our analytical procedures. Static delivery of CO<sub>2</sub> at rates that turned over experimental vessels several times an hour resulted in constant pH levels during experiments [ $<0.5\%$  relative standard deviation (RSD) within treatments based on multiple daily measurements]. The rates of larval growth, development, and survivorship and lipid content of larvae were monitored through metamorphosis. Differences in the sizes and shells of larvae and early juvenile stage individuals were documented by cross-sectioning individuals and observing them with an SEM.

## Results and Discussion

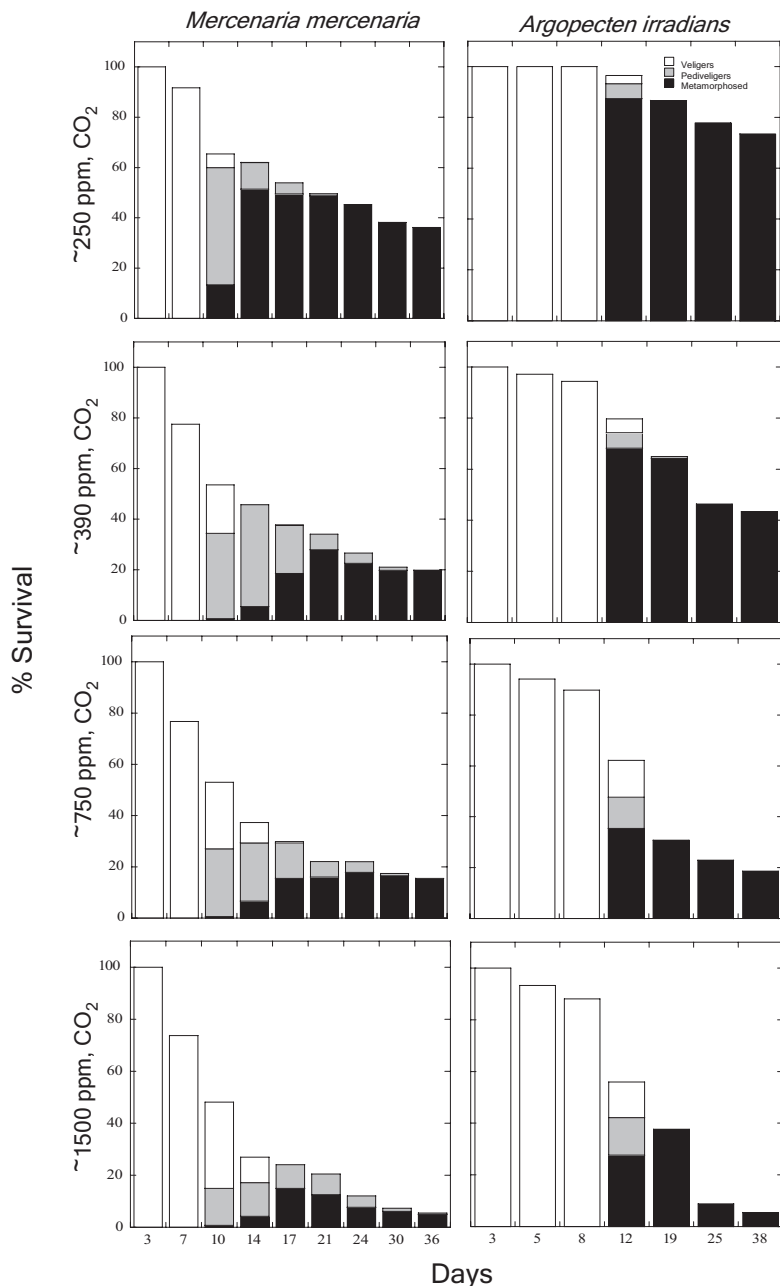
Larvae grown under near preindustrial levels of CO<sub>2</sub> (250 ppm) displayed the highest rates of metamorphosis, growth, and survival. After 36 d of development, 40% of *M. mercenaria* grown under ~250 ppm CO<sub>2</sub> had survived, whereas only 20% survived at modern day CO<sub>2</sub> levels (~390 ppm), and only 6% survived at ~1,500 ppm CO<sub>2</sub> ( $P < 0.001$ ; Fig. 1). *A. irradians* displayed similar patterns, with 74% of individuals surviving 38 d under ~250 ppm CO<sub>2</sub>, 43% surviving at ~390 ppm, and only 5.4% remaining at ~1,500 ppm CO<sub>2</sub> ( $P < 0.001$ ; Fig. 1). Larvae grown under the lowest CO<sub>2</sub> levels displayed remarkably faster rates of metamorphosis compared with individuals grown under present day CO<sub>2</sub>. For example, after 14 d of development, 51% of *M. mercenaria* larvae had fully metamorphosed at ~250 ppm CO<sub>2</sub>, whereas <7% had done so under higher levels of CO<sub>2</sub> ( $P < 0.001$ ; Fig. 1). After 12 d of development, *A. irradians* larvae displayed a somewhat similar trend because 87% of the larvae had metamorphosed at ~250 ppm CO<sub>2</sub>, whereas 68% had done so at ~390 ppm CO<sub>2</sub> ( $P < 0.001$ ; Fig. 1). The mean diameters attained by both species of larvae also were strongly affected by CO<sub>2</sub>. *M. mercenaria* and *A. irradians* larvae grown under 250 ppm CO<sub>2</sub> (523 ± 38 and 531 ± 51 μm) were significantly larger than those grown under present day (282 ± 5 and 449 ± 35 μm) and higher (210 ± 9 and 311 ± 26 μm at ~1,500 ppm) levels of CO<sub>2</sub> ( $P < 0.001$ ; Fig. 2). These trends in the size of individuals were obvious during the examination of individuals under SEM (Figs. 3 and 4).

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The authors declare no conflict of interest.

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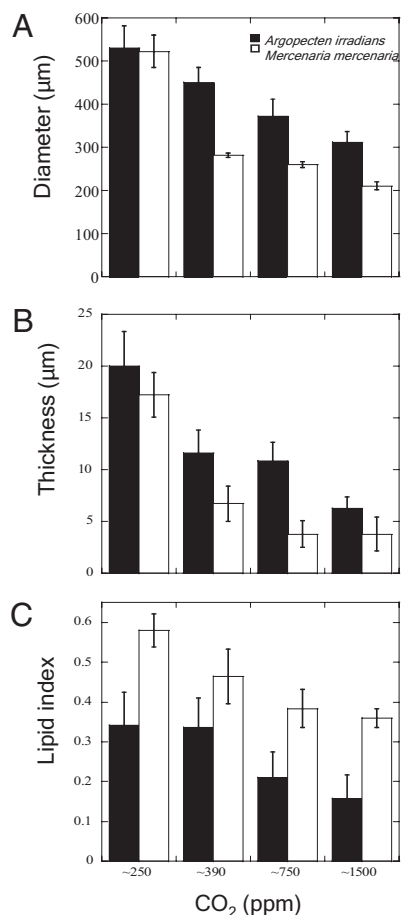
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**Fig. 1.** Development and survival of *M. mercenaria* and *A. irradians* larvae. Percent survival and developmental stage (veliger, pediveliger, and metamorphosed) of larvae grown under four levels of CO<sub>2</sub>, ~250, 390, 750, and 1,500 ppm (Table 1). The relative SD of larval survival among replicated vessels per treatment for all times points and experiments was 4% ( $n = 4$  per treatment).

Levels of CO<sub>2</sub> strongly influenced the early formation of *M. mercenaria* and *A. irradians* shells. For example, after 17 d of development *M. mercenaria* shells were  $17 \pm 2$   $\mu\text{m}$  thick under ~250 ppm CO<sub>2</sub>,  $6.7 \pm 2$   $\mu\text{m}$  at ~390 ppm CO<sub>2</sub>, and  $3.8 \pm 1$   $\mu\text{m}$  at ~750 ppm and ~1,500 ppm CO<sub>2</sub> ( $P < 0.001$ ; Figs. 2 and 3). *A. irradians* shells also decreased in thickness with increasing CO<sub>2</sub> being  $20 \pm 3$ ,  $12 \pm 1$ ,  $11 \pm 1$ , and  $6.3 \pm 1$   $\mu\text{m}$  thick under ~250, 390, 750, and 1,500 ppm CO<sub>2</sub> ( $P < 0.001$ , Fig. 2), respectively. Beyond impacting shell thickness, elevated levels of CO<sub>2</sub> severely altered the development of the hinge structure of early stage bivalves. As CO<sub>2</sub> levels increased from ~250 to ~1,500 ppm, there were dramatic declines in the size, integrity, and connectedness of the hinge (Fig. 3). Although the *M. mercenaria* hinge displayed a “tongue and groove” pattern under low CO<sub>2</sub> (250 and 390 ppm), under higher CO<sub>2</sub>

concentrations the hinge and associated hinge teeth became increasingly separated and detached. Given that the bivalve hinge facilitates opening and closing of shells, allowing for intake of food and the excretion of waste (18), the compromised hinges observed under elevated CO<sub>2</sub> may hinder the ability of individuals to obtain and process suspended particles for nutrition. This hypothesis is consistent with changes in lipid stores of larval shellfish exposed to differing CO<sub>2</sub> concentrations. For both species, with each increasing level of CO<sub>2</sub>, the lipid content (as estimated by an index) decreased significantly ( $P < 0.001$ ; Fig. 2). Increasing CO<sub>2</sub> concentrations also caused marked changes in the morphology of the outer edge of juvenile shells (Figs. 3 and 4). With increasing levels of CO<sub>2</sub>, this region of the shell became increasingly riddled with holes, pockmarks, and crevices, observations consistent with other



**Fig. 2.** Diameters, shell thickness, and lipid index of bivalve larvae grown under a range of CO<sub>2</sub> concentrations. Data are from four levels of CO<sub>2</sub>, ~250, 390, 750, and 1,500 ppm. (A) Diameters of *M. mercenaria* (day 24) and *A. irradians* (day 20). (B) Thickness of *M. mercenaria* (day 36) and *A. irradians* (day 52) shells at midpoint between the hinge and valve edge of the upper and lower shell of cross sectioned individuals. (C) Lipid index (lipid area/total area) for *M. mercenaria* (day 24) and *A. irradians* (day 20). Error bars represent SD of replicated vessels per treatment ( $n = 4$  per treatment).

juvenile and larval shellfish reared under high CO<sub>2</sub> (10, 14), suggesting CaCO<sub>3</sub> shells were malforming and/or dissolving under more acidic conditions. Altered shell morphology was also obvious in juvenile scallops that had distinct ridges, characteristic of later stages of development, under preindustrial CO<sub>2</sub>, whereas individuals reared under higher CO<sub>2</sub> conditions lacked ridges, a sign of slower development (Fig. 4 and 19).

Shell integrity is one of the most important lines of defense for larval and juvenile bivalve shellfish, because shells provide physical support for soft and delicate internal organs (20) and protection from benthic and pelagic predators and suspended particles (21, 22). As such, the thinner, frailer shells displayed by early life history bivalves reared under modern day and elevated CO<sub>2</sub> would likely make individuals more vulnerable to predation and/or other environmental stressors. Similarly, within an ecosystem setting, larvae that accumulate fewer lipids (Fig. 2) are generally slower to metamorphose (19) and are more likely to perish once settled (23). Finally, individuals with extended metamorphosis times (Fig. 1) and that are smaller (Fig. 2) would be susceptible to greater rates of predation and natural mortality (23, 24). Hence, within an ecosystem setting, mortality rates of early life history bivalves that develop under modern day and higher CO<sub>2</sub> levels would be expected to be even greater than the rates observed during our experiments. Given

that bivalves in coastal areas naturally experience extremely high mortality rates in the transition from larvae to benthic juveniles (9), increases in mortality due to elevated CO<sub>2</sub> could have profound effects on estuarine bivalve populations (5).

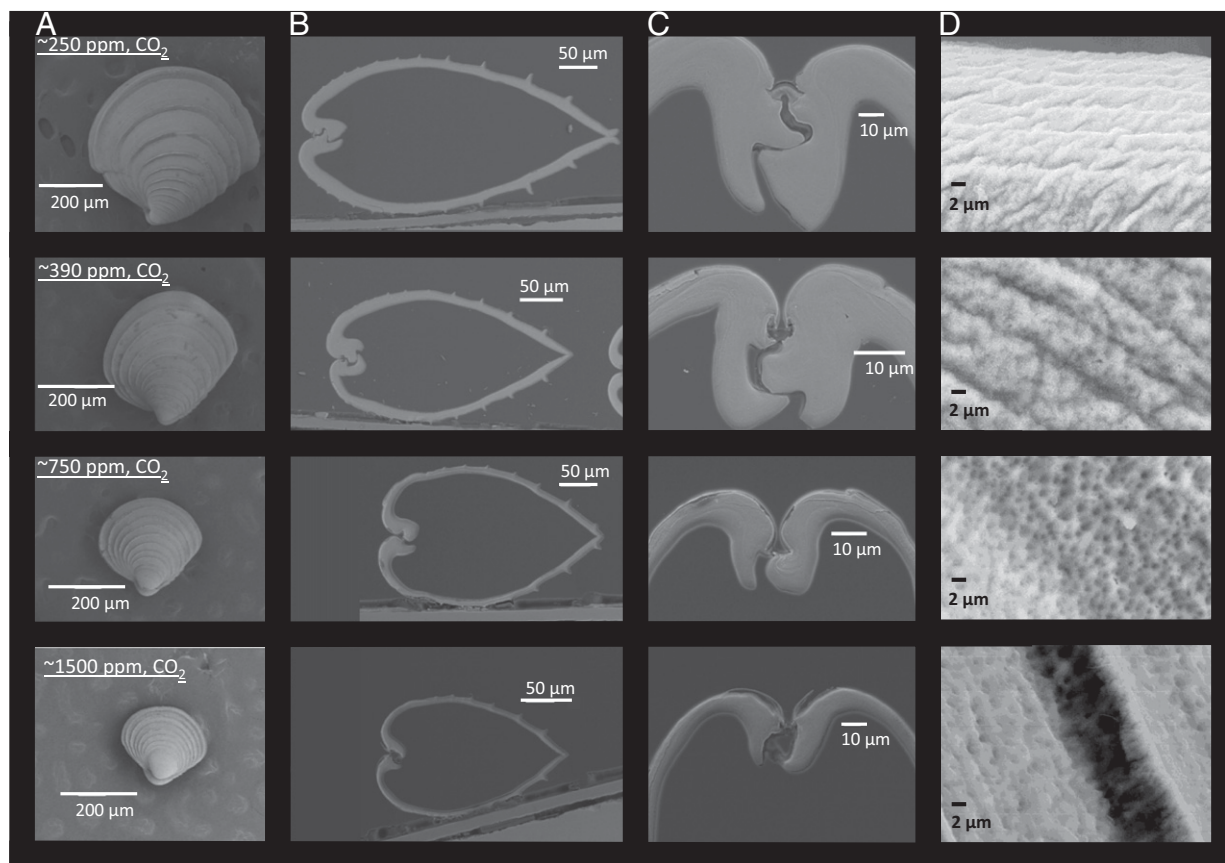
Our findings regarding the effects of future CO<sub>2</sub> levels on larval shellfish are consistent with recent investigations of ocean acidification demonstrating that calcifying organisms will experience declines in survival and growth, as well as malformed CaCO<sub>3</sub> shells and hard parts (25). However, our examination of the development of larval shellfish at levels of CO<sub>2</sub> present before the industrialization of the planet provides important insight regarding the potential effects ocean acidification has had on calcifying organisms during the past two hundred years. Consistent with our findings, larval oysters (*Crassostrea virginica*) have displayed slightly larger shell area when grown under preindustrial CO<sub>2</sub> levels compared with modern levels (26).

During the ~24 million years before the industrial revolution, atmospheric CO<sub>2</sub> levels are estimated to have been relatively static, likely fluctuating in a narrow range significantly below the concentrations present today (27, 28). Moreover, periods of higher CO<sub>2</sub> before this era may not have been accompanied by lower pH and carbonate ion concentrations because the oceans may have buffered the more gradual changes in CO<sub>2</sub> that have occurred through geological history (3, 29). The evolution of calcification in ocean animals is unknown, and the multiple forms of CaCO<sub>3</sub> synthesized by modern day calcifiers (calcite, aragonite, amorphous CaCO<sub>3</sub>, and high magnesium CaCO<sub>3</sub>) differ widely in their vulnerabilities to dissolution under lower pH (30). Although the precise evolutionary tracks of modern bivalves remain somewhat uncertain (31), fossil evidence suggests that 906 of the 958 living genera of bivalve mollusks, including the species presented here, have a record that began in the mid- to late Cenozoic with the greatest continuous increase in genera between ~15 and ~25 Mya (32), a period of estimated lower CO<sub>2</sub> levels compared with today (27, 28). Together with our results, this suggests that ocean acidification since the industrial revolution may have applied selection pressure on modern marine bivalves and may continue to do so in the future.

The shallow marine environments that many marine bivalves occupy can harbor dynamic levels of pH and CO<sub>2</sub> (33, 34) and the precise degree of phenotypic plasticity of survival among bivalve larvae in the face of higher CO<sub>2</sub> has not been established. Adaptation and evolution could promote the proliferation of bivalve strains that are more resistant to the increases in ocean CO<sub>2</sub> expected in the coming century and some calcifying organisms may even benefit from higher CO<sub>2</sub> levels (25, 35). Importantly, however, the current rates of increase in atmospheric CO<sub>2</sub> are significantly faster than any recorded in tens of millions of years (27, 28), suggesting this evolutionary challenge may be without precedent for extant calcifying species.

A comparison of our two study species may provide insight into future evolutionary pressure of ocean acidification on marine calcifiers. Globally, *M. mercenaria* has a larger, more diverse geographic distribution (36) than *A. irradians* (37), an attribute that generally provides resistance to evolutionary pressures (38) such as increasing CO<sub>2</sub> levels. In addition, predicted extinction rates are higher for the marine mollusk family Pectinidae, which includes *A. irradians*, than the Veneridae family, which includes *M. mercenaria* (39). This information, combined with the more dramatic declines in survival displayed by *A. irradians* under higher CO<sub>2</sub> levels compared with *M. mercenaria* (Fig. 1), suggests *A. irradians* may face a greater evolutionary challenge in adapting to future increases in CO<sub>2</sub> concentrations.

Precipitous declines in wild populations of bivalves during the 20th century have been attributed to overfishing, loss of habitat, hypoxia, and harmful algal blooms (40, 41). Our results suggest that ocean acidification is another process that may have contributed to the declines of these populations in the recent past



**Fig. 3.** SEM images of *M. mercenaria* larvae grown under a range of CO<sub>2</sub> concentrations. SEM images of 36-d-old *M. mercenaria* grown under different levels of CO<sub>2</sub>, ~250, 390, 750, and 1,500 ppm (Table 1). (A) Images of individual larvae under each CO<sub>2</sub> level. (B) Hinge to valve cross sections of individuals under each CO<sub>2</sub> level. (C) The hinge of individuals under each CO<sub>2</sub> level. (D) A magnification of the outermost shell of individuals under each CO<sub>2</sub> level.

and could further impact bivalve population densities and diversity in the future. Looking forward, marine organisms will be threatened by aspects of climate change beyond elevated CO<sub>2</sub>, including higher temperatures. Given that the rise in ocean temperatures projected for the coming century (4) is within a range that could also hinder the growth and survival of bivalve larvae (19, 42), future studies should consider the impact of higher CO<sub>2</sub> in conjunction with temperature changes in line with such projections.

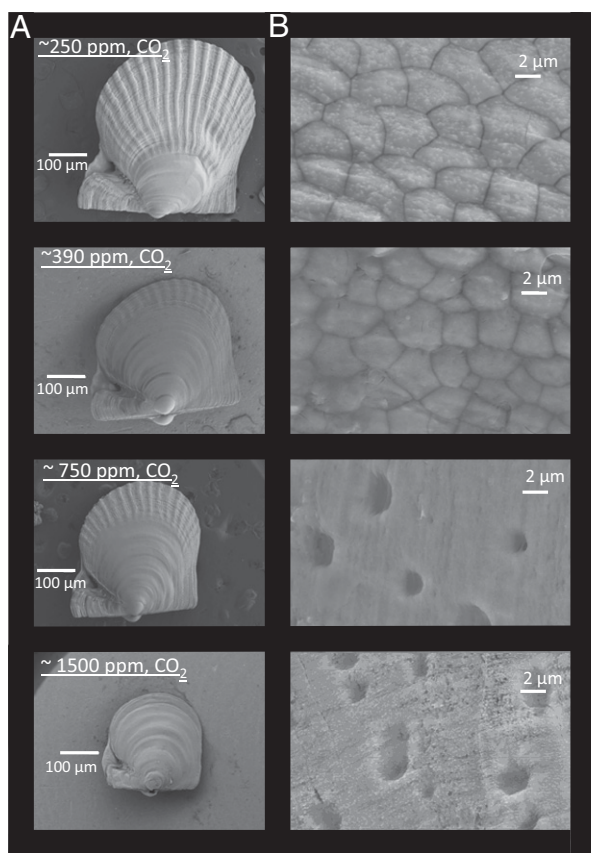
## Materials and Methods

**CO<sub>2</sub> Treatments and Measurements.** A gas proportionator system (Cole Parmer Flowmeter system, multitube frame) was used to deliver CO<sub>2</sub> gas to seawater treatments at multiple rates. The gas proportionator mixed appropriate flow rates of 5% carbon dioxide gas, low carbon dioxide gas, and pressurized air (~390 ppm CO<sub>2</sub>) to yield the concentrations of carbon dioxide desired for experiments at a net flow rate ( $350 \pm 5$  mL min<sup>-1</sup>) that turned over the volume of plexiglass covered experimental beakers >400 times daily. Experiments were repeated with tanked gas premixed at each specific CO<sub>2</sub> level and nearly identical seawater chemistry and larval responses were obtained. For experiments, the CO<sub>2</sub> gas mixtures from the proportionator system were continuously delivered to the bottom of four replicated, polypropylene 1-L beakers containing 0.2 μm filtered seawater from eastern Shinnecock Bay, NY. With continuous bubbling, all treatment beakers remained saturated with respect to oxygen (~8 mg L<sup>-1</sup>). To quantify precise CO<sub>2</sub> levels attained in experimental beakers, seawater in beakers was bubbled for 24 h and analyzed at the start (immediately before the addition of larvae and phytoplankton) and at the end (larvae removed, phytoplankton present) of each experiment using an EGM-4 Environmental Gas Analyzer (PP Systems) system that quantifies total dissolved inorganic carbon levels after separating the gas phase from seawater using a Liqui-Cel Membrane (Membrana). This instrument provided a methodological precision  $\pm 3.6\%$

for replicated measurements of total dissolved inorganic carbon and provided full recovery ( $102 \pm 3\%$ ) of Dr. Andrew Dickson's (Scripps Institution of Oceanography, University of California at San Diego, La Jolla, CA) certified reference material for total inorganic carbon in seawater (batch 102 = 2,013 μmol DIC kg seawater<sup>-1</sup>). Levels of CO<sub>2</sub> were subsequently calculated based on measured levels of total inorganic carbon, pH (total scale; mol kg seawater<sup>-1</sup>), temperature (~24 °C), salinity (~28 ppt), and first and second dissociation constants of carbonic acid in seawater according to Roy et al. (43) using the program CO2SYS (<http://cdiac.ornl.gov/ftp/co2sys/>). Multiple daily measurements of pH (calibrated prior each use with NIST traceable standards,  $\pm 0.002$ , Orion Star Series Benchtop pH meter; Thermo Scientific) indicated experiment beakers maintained a constant pH level throughout all experiments (<0.5% RSD within treatments).

**Experimental Design.** The drafted recommendations of the "best practices" for small microcosm experiments set forth by European Project on Ocean Acidification (EPOCA) were followed for this project. For example, aeration of seawater was used to reach a target pCO<sub>2</sub> level, the ideal mechanism to manipulate seawater carbon chemistry (43). Experiments were conducted using two species of bivalves: *M. mercenaria* and *A. irradians*. For each experiment, four levels of carbon dioxide were administered: a high level (~1,500 ppm CO<sub>2</sub>), predicted for the year 2250; a moderate level (~750 ppm CO<sub>2</sub>), predicted for the year 2100 (3, 44); ambient air (~390 ppm CO<sub>2</sub>); and a near preindustrial level (~250 ppm CO<sub>2</sub>; 25, 26 and Table 1). *A. irradians* and *M. mercenaria* larvae were obtained from locally obtained broodstock spawned at the East Hampton Shellfish Hatchery.

A culture of *Isochrysis galbana* (Tahitian strain, T-Iso) was maintained in exponential phase growth using standard culture conditions and added at a density of  $2 \times 10^7$  cells daily to each experimental beaker ( $2 \times 10^4$  mL<sup>-1</sup>) as a food source. This algal species administered at this density and at this rate is known to produce high growth rates and survivorship of shellfish larvae through metamorphosis (19, 42, 45). To promote the high survivorship, containers that were in contact with larvae were never exposed to chemicals



**Fig. 4.** SEM images of 52-d-old *A. irradians* grown under different levels of  $\text{CO}_2$ : ~250, 390, 750, and 1,500 ppm, (Table 1). (A) Image of a full individual larvae under each  $\text{CO}_2$  level. (B) A magnification of the outermost shell of individuals under each  $\text{CO}_2$  level.

or detergents (45). To discourage the growth of bacteria during experiments, an antibiotic solution (5,000 units of penicillin, 5 mg of streptomycin, and 10 mg of neomycin per mL of solution, No.4083; Sigma-Aldrich) was added to each beaker at 1% its original concentration at the beginning of

each experiment and during each water change (approximately two times weekly). This antibiotic mixture at this concentration has been shown to have no negative effects on the growth and survivorship of shellfish larvae (45). For each experiment, ~200 larvae were distributed to each experimental beaker, achieving an environmentally realistic abundance of larvae (42). Each treatment began with ~900 mL to allow enough beaker volume for the algal culture to be added daily as a food source. Twice weekly during experiments, larvae were gently poured onto a 64- $\mu\text{m}$  mesh, and the condition (live or dead) and developmental stage of each larvae (veligers, pediveligers, and metamorphosed) was determined visually under a dissecting microscope; every individual larvae was counted at every water change. Larvae from each beaker ( $n = 4$ , per treatment) were removed, counted, observed, and transferred into a new beaker with new filtered seawater, food, and antibiotics within a 15 min period. Throughout experiments, all beakers were submerged in a water bath maintained at 24 °C via the use of commercially available heaters and chillers. This temperature generally yields high growth rates for *A. irradians* and *M. mercenaria* larvae (19, 42). Percent survivorship of all larvae was determined at each of the biweekly water changes when the numbers of larvae in each stage of veligers, pediveligers, and metamorphosed juveniles were quantified. Experiments were terminated after all surviving larvae in all treatments had metamorphosed. To statistically evaluate the effect of  $\text{CO}_2$  treatments on larval survival, goodness of fit tests (G Tests) were performed (46).

**SEM.** To document differences in the size and structure of larval and early juvenile shellfish exposed to differing levels of  $\text{CO}_2$ , randomly chosen individuals ( $n = 4$  per treatment) were mounted for SEM in two distinct ways. Firstly, to image the outside of shells, individuals were attached at 45° relative to a level surface to a conductive substrate using carbon, double-sided tape and were subsequently coated with ~12 nm of gold using an Edwards 150B rotary pump. To image the thickness and internal dimensions, cross-sections of shellfish were prepared. Individuals were mounted on glass microscope slides using UV-curing adhesive coating (Locite 4304) and were impregnated with low-viscosity epoxy (Stuers' Specifix-20) under vacuum outgassing, a step that did not alter the original shape or size of individuals. After curing, the epoxy mount was progressively ground and polished to the centerline (hinge to shell edge) of the shellfish using silicon carbide sandpapers, followed by successively finer diamond polishing grits (15, 6, and 3  $\mu\text{m}$ ), 0.05  $\mu\text{m}$  aluminum oxide suspension, and finally with colloidal silica. All individuals were cross-sectioned at the same location (hinge to shell edge) across the shell. This mount was then attached to a conductive substrate using carbon double-sided tape and coated with ~4 nm of gold. SEM images were collected on both types of samples with a Leo (Zeiss) Model #1550 electron microscope using a high voltage of 20 KV and a Robinson backscatter detector. All components of individual bivalve shells displayed in Figs. 3 and 4 were probed using advanced EDAX/EDA microanalysis in the LEO

**Table 1.** Temperature, pH, carbonate chemistry, alkalinity, and salinity ( $\pm\text{SD}$ ) during the four-level  $\text{CO}_2$  experiments with *M. mercenaria*, and *A. irradians* larvae

Parameter	Near preindustrial $\text{CO}_2$	Ambient, present day $\text{CO}_2$	Year 2100 $\text{CO}_2$	Year 2200 $\text{CO}_2$
<i>M. mercenaria</i>				
Temperature (°C)	24 $\pm$ 0.52	24 $\pm$ 0.52	24 $\pm$ 0.52	24 $\pm$ 0.52
pH	8.171 $\pm$ 0.022	8.052 $\pm$ 0.036	7.801 $\pm$ 0.004	7.532 $\pm$ 0.021
p $\text{CO}_2$ (ppm)	247.1 $\pm$ 6.231	380.0 $\pm$ 33.02	742.3 $\pm$ 9.111	1516 $\pm$ 31.21
$\Omega_{\text{calcite}}$	5.31 $\pm$ 0.47	4.53 $\pm$ 0.41	2.82 $\pm$ 0.05	1.67 $\pm$ 0.05
$\Omega_{\text{aragonite}}$	3.42 $\pm$ 0.30	2.92 $\pm$ 0.26	1.82 $\pm$ 0.03	1.08 $\pm$ 0.03
Total DIC ( $\mu\text{mol L}^{-1}$ )	1646 $\pm$ 94.21	1831 $\pm$ 52.34	1947 $\pm$ 21.33	2108 $\pm$ 18.06
$\text{CO}_3^{2-}$ ( $\mu\text{mol L}^{-1}$ )	208.0 $\pm$ 20.22	178.0 $\pm$ 16.03	111.0 $\pm$ 1.806	66.0 $\pm$ 1.904
Alkalinity (TA)	1938 $\pm$ 117.3	2070 $\pm$ 66.42	2080 $\pm$ 22.63	2127 $\pm$ 49.71
Salinity	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0
<i>A. irradians</i>				
Temperature (°C)	24 $\pm$ 0.51	24 $\pm$ 0.52	24 $\pm$ 0.52	24 $\pm$ 0.52
pH	8.170 $\pm$ 0.026	8.041 $\pm$ 0.044	7.801 $\pm$ 0.005	7.530 $\pm$ 0.011
p $\text{CO}_2$ (ppm)	244.1 $\pm$ 4.006	386.5 $\pm$ 40.04	738.9 $\pm$ 9.941	1529 $\pm$ 35.05
$\Omega_{\text{calcite}}$	5.18 $\pm$ 0.06	4.55 $\pm$ 0.47	2.81 $\pm$ 0.06	1.66 $\pm$ 0.05
$\Omega_{\text{aragonite}}$	3.34 $\pm$ 0.35	2.94 $\pm$ 0.30	1.81 $\pm$ 0.04	1.07 $\pm$ 0.03
Total DIC ( $\mu\text{mol L}^{-1}$ )	1613 $\pm$ 53.54	1850 $\pm$ 30.98	1941 $\pm$ 25.54	2101 $\pm$ 9.221
$\text{CO}_3^{2-}$ ( $\mu\text{mol L}^{-1}$ )	202.0 $\pm$ 23.42	180.0 $\pm$ 18.44	111.0 $\pm$ 2.341	66.02 $\pm$ 1.911
Alkalinity (TA)	1899 $\pm$ 35.24	2090 $\pm$ 50.01	2075 $\pm$ 26.84	2146 $\pm$ 11.21
Salinity	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0	28.0 $\pm$ 1.0

(Zeiss) Model #1550 electron microscope and were confirmed to contain almost exclusively C, O, and Ca.

**Size and Lipid Analysis.** To estimate the relative lipid content of larvae, Nile Red stain was used to bind to neutral lipids and fluoresce under an FITC filter on an epifluorescent microscope (23, 47). A Nile Red stock solution was made of 1.25 mg of Nile Red crystals in 100 mL of acetone. Randomly selected larvae ( $n = 15$ ) from each treatment were stained with a 1:9 dilution of the stock solution and 0.2  $\mu\text{m}$  filtered seawater. Larvae were exposed to the stain for  $\sim 1.5$  h, rinsed with filtered seawater, and digitally photographed with a Roper Scientific Photometrics CoolSNAP ES camera under an epifluorescent microscope. Digital images of each larva were analyzed for the

area of lipid accumulation and the diameter and the area of individuals using ImageJ. A lipid index was estimated by dividing the area of the larvae containing the fluorescing lipids by the total larval area, thereby allowing for direct comparisons among treatments. One-way ANOVAs and posthoc Tukey multiple comparison tests were performed to examine the differences among larval lipid indexes, shell length, and thickness, at each  $\text{CO}_2$  level.

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